8.

Name		***************************************
Par.	Nr.	****************

THIRD SEMESTER B.Sc. DEGREE EXAMINATION, NOVEMBER 2017

(CUCBCSS-UG)

Complementery Course

	Complement	ary Course	
CH	E 3C 03—ORGA	NIC CHEMISTRY	
Time: Three Hours			Maximum : 64 Marks
•	Section A (One w	vord/Sentence)	- ALLIA ILD
	Answer all the Each carries		
1. Higher homologues of Alka	ane can be prepare	d by the reaction:	
(i) Wortz.	(ii)) HVZ.	
(iii) Hydrogenation.	(iv)	Haloform.	
2. How many stereoisomers ar	e possible in Lactic	acid?	
3. E-Z designation is used in t	he case of:	n en	
(i) Geometrical isomeri	sm. (ii)	Optical isomerism.	
(iii) Position isomerism.	(iv)	Functional group isome	erism.
4. According to Huckle rule, w	hat is the value of		
5. When - OH, - COOH, - CN group is:			
(i) - OH.	(ii)	$-NO_2$.	
(iii) – COOH.	(iv)	– CN.	•
6. Which is the terpenoid occur	in peppermint oil	?	
7. Pick out the Polysaccaride from	· · · · · · · · · · · · · · · · · · ·	•	
(i) Sucrose.	(ii)	Maltose.	
(iii) Starch.	(iv)	Lactose.	
3. Out of the following which alk	cene is more stabl		
(i) Cis-2 butene.	(ii)	Trans-2 butene.	
(iii) Propene.	(iv)		
. How many Pi electron present	•	Ethene.	
Homolytic fission of covalent b	ond liberates —		
	· •		$(10 \times 1 = 10 \text{ mark})$

Turn over

Section B (Short Answers)

Answer any seven questions.

Each carries 2 marks.

- 11. Explain Kolbe's electrolytic method.
- 12. Explain why chair conformation of Cyclohexane is more stable than boat conformation.
- 13. Explain Huckle rule's of aromaticity.
- 14. How many optical isomers possible for the compound 3-chloro 2-butanol? Draw the structure.
- 15. Phenol is stronger acid than an alcohol. Why?
- 16. Briefly describe the structure of starch.
- 17. Compare the stability of 1-butene and 2-butene.
- 18. Explain the process vulccanisation.
- 19. What are fibrous proteins?
- 20. Write the effect of structure on reactivity of Alkyl halide by SN² mechanism.

 $(7 \times 2 = 14 \text{ marks})$

Section C (Paragraphs)

Answer any four questions.

Each carries 5 marks.

- Write a brief note on the optical activity of Tataric acid.
- Write the method of preparation of Methyl orange and explain its uses. 22.
- What is meant by Hydrogenation of oils and explain its application? 23.
- 24. Discuss the double helical structure of DNA.
- Discuss the hybridisation and stability of Carbocations. **25**. •
- Explain Lucas test for differentiating 1°, 2° and 3° structure of alcohols. **26**.

 $(4 \times 5 = 20 \text{ mark})$

Section D (Essays)

Answer any two questions.

Each carries 10 marks.

- 27. Discuss the Primary, Secondary and Teritiary structure of proteins.
- 28. Write a brief note on:
 - (i) Enantiomers.

(ii) Meso compounds.

(iii) Chirality.

- Conformations of Ethane. (iv)
- (i) Give one method of preparation of 3° Alcohol by using Grignard reagent. 29.
 - (ii) Explain Haloform reaction.
 - (iii) Compare the acidity of Phenol, P-nitrophenol and Para-methoxy phenol.
- 30. Write a notes on:
 - Industrial application of cellulose.
 - Classification of terpenes with examples. (ii)
 - (iii) Physiological functions of Nicotine.

 $(2 \times 10 = 20 \text{ m})$