The HUPAG name of

FOURTH SEMESTER B.Sc. DEGREE EXAMINATION, MAY 2011

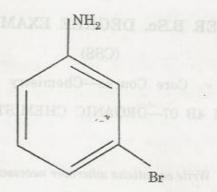
(CSS)

Core Course—Chemistry

CH 4B 07-ORGANIC CHEMISTRY-I

lime : Three Hours			Maximum Weightage: 30
	ons where	ver necessary.	
I. Multiple choice and fill in the blanks ty	pe questi	ons. Answer all to	welve questions:
1 Corey-House reaction is governed			
(a) Electrophilic substitution.			
(b) Nucleophilic substitution.			
(c) Free radical substitution.			
(d) Polymerisation reaction.			
2 The most stable conformation of n	-butane is	s the ——— cor	nformation.
(a) eclipsed.	(b)	staggered.	
(c) skew.	(d)	anti.	
3 Which alkene among the following	g is most s	stable?	
(a) Cis-2-butene.	(b)	Trans-2-butene	
(c) 1.butene.	(d)	all are equally s	table.
4 Kharasch effect is observed in the	addition	of ——— to an	unsymmetrical alkene.
(a) HBr.	(b)	HCl.	
(c) HI.	(d)	all of the above.	a studenda selvenima A BI
5 A reagent for cis hydroxylation is	:		
(a) OsO4.		H ₂ O ₂ /H ⁺ .	
(c) benzoyl peroxide.	(d)	none of the abo	ve.
6 is a neutral electrophile.			
- m I for the exidetic		ne to benzoic acid	l is——.
m 1 Colombia is			
8 The shape of carbocation is			

- 9 Meso tartaric acid is optically inactive due to ------ compensation.
- 10 The IUPAC name of:

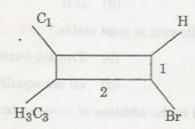


is -----

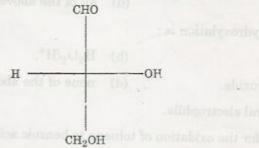
- 11 Citral is used for -----.
- 12 PMMA is -----.

 $(12 \times \frac{1}{4} = 3)$

- II. Short answer type questions. Answer all nine questions:
 - 13 What is Wurtz reaction?
 - 14 Draw the most stable confrontation of methylcyclohexane.
 - 15 What happens when ethyne is passed over ammoniacal silver nitrate?
 - 16 Name the following alkene: -



- 17 What are nitrenes?
- 18 Assign the absolute configuration of the following molecule: -



- 19 Draw an optically active compound, does not having a chiral carbon.
- 20 Write the correct order of stability of carbanions.
- 21 Draw the structure of Limonene.

- III. Short essays or paragraph questions. Answer any five questions:
 - 22 Illustrate asymmetric synthesis citing an example.
 - 23 Explain the Haworth's synthesis of naphthalene.
 - 24 Give a brief account of the structure of natural rubber.
 - 25 Discuss briefly the importance of the following reactions: -
 - (a) Ozonolysis.
 - (b) Periodic acid oxidation.
 - 26 Explain the structure and shape of ethylene molecule.
 - 27 Give a brief note on ring strains in cyclopropane and cyclobutane.
 - 28 Explain the mechanism of addition of HBr to propene in the presence and absence of Peroxides.

 $(5 \times 2 = 10)$

- IV. Essay questions. Answer any two questions:
 - 29 Write notes on optical isomerism, methods of resolution and Optical activity of Biphenyls.
 - 30 Discuss the mechanisms of nitration and sulphonation of naphthalene and bromination of benzene.
 - 31 Discuss the structure, hybridisation and stability of carbocations and carbenes.

 $(2 \times 4 = 8)$