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FIFTH SEMESTER B.Sc. DEGREE EXAMINATION NOVEMBER 2011

(CCSS)

Chemistry - Core Course - VI

CH 5B 10 - ORGANIC CHEMISTRY - II

		CH 5B 10 - ORGAL	MIC CHEMIS	1111-11
ro	e Hours			Maximum Weightage: 30
11 6	110010	Write equations	wherever nece	essary.
ult	iple choice an	d fill in the blanks type q	uestions. Ans	swer all twelve questions:
		est reagent for allylic bro		
		NBS.		Bromine water.
	(c)	Bromine in CCl ₄ .	(d)	All of the above.
2.	Which one of	the following halide is m	ost reactive t	owards nucleophilic substitution?
	(a)	Chorobenzene.		Iodobenzene.
	(c)	Vinyl bromide.	(d)	Allyl bromide.
3.	Which metal-	-C bond is most ionic?		
	(a)	C-Mg.	(b)	C-Zn.
	(c)	C-Li.	(d)	All are equally ionic.
4.	Grignard rea	gent reacts with a ketone	e followed by	hydrolysis gives
	A. Carrier	Primary alcohol.	(b)	Tertiary alcohol.
	(c)	Secondary alcohol.	(d)	A mixture of the above.
5.	Phenol is a s	tronger acid than	On Park Said	
	(a)	Formic acid.	(b)	o-Nitrophenol.
	(c)	p-Nitrophenol.	(d)	o-Cresol.
6.	Williamson's	synthesis of ether is an	example of _	substitution.
		Nucleophilic.	(b)	Electrophilic.
		Free radical.		None of the above.
7.	Preparation	of ethers by alkoxy-merc	curation invol	lves as a reducing agent in the
	final stage.			
	(a)	${ m LiAlH_4}.$	(b)	Na BH ₄ .
	(c)	Ni-H ₂ .	(d)	Pd-H ₂ .
				m

8.	8. The nucleophilic addition reactions on carbonyl groups are catalysed by			
	(a)		(b)	
	(c)	Acids.	(d)	Ampholytes.
9.	Methyl keto	nes are easily identified by	d'av	0/4
	(a)	Iodoform reaction.	(b)	Schiff's test.
	(c)	Fehling's test.	(d)	Tollen's test.
10.	The natural	source for formic acid is		
	(a)	Vinegar.	(b)	Red ant.
	(c)	Butter.	(d)	Valerian plant.
11.	Phthalic acid	l reacts with resorcinol in preser	nce of	Con., H ₂ SO, gives:
	(a)	Phenolphthalein.	(b)	Alizarin,
	(c)	Coumarin.	(d)	Fluorescein.
12.	Pericyclic rea	actions generally involves :		
	(a)	Carbocations.	(b)	Carbanions.
	(c)	Free radicals.	(d)	No intermediates.
				$(12 \times \frac{1}{4} = 3 \text{ weight})$
Short	Answer Type	Questions. Answer all nine que	estions	

II. S

- 13. What are electrocyclic reactions? Give an example.
- 14. Explain Kolbe's reaction.
- 15. Explain any two applications of crown ethers in organic synthesis.
- 16. How is iodoform prepared from ethanol?
- 17. Which is more reactive and why? Ethanol or acetone.
- 18. What happens when cinnamic acid is exposed to sunlight?
- 19. Amides are very slowly hydrolysed by water? Why?
- 20. Give an example of a pericyclic reaction taking place in human body.
- 21. Vinyl halides are aliphatic halogen compounds but they resemble aryl halides in chemical reactions. Rationalise.

 $(9 \times 1 = 9 \text{ weights})$

III. Short Essays or Paragraph Questions. Answer any five questions:

- 22. Explain the benzyne intermediate mechanism of nucleophilic aromatic substitution furnish any evidence in support of the mechanism.
- 23. Discuss the mechanism of Reformatsky reaction. Mention any one of its synthetic application.

- 24. Discuss in detail Zeisel's method of methoxy groups.
- 25. Explain the chemistry of Liebermann's nitroso reaction.
- 26. Discuss Diels-Alder reaction using FMO method.
- 27. Explain the mechanism of Claisen rearrangement.
- 28. How is phenolphthalein prepared? Why is it colourless in strong alkali?

 $(5 \times 2 = 10 \text{ weightage})$

Essay Questions. Answer any two questions:

- 29. Discuss the effect of the structure of the substrate and polarity of the solvent on $S_N 1$ and $S_N 2$ reactions.
- 30. Discuss the mechanisms of the following reactions:
 - (a) Aldol condensation.
 - (b) Cannizaro's reaction.
 - (c) Claisen condensation.
 - (d) Benzoin condensation.
- Give a detailed account of the effects of substituents on the acidity of aliphatic and aromatic carboxylic acids.

 $(2 \times 4 = 8 \text{ weightage})$