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Maximum: 30 Weightage

## FIFTH SEMESTER B.Sc. DEGREE EXAMINATION, OCTOBER 2012

(CCSS)

Chemistry

## CH 5B 10—ORGANIC CHEMISTRY—II

(Core Course)

Ihr	ree Hour	S		
		Write	equations where	ever necessary.
L M	ultiple c	hoice and fill in the bl	anks type quest	ions. Answer all twelve questions:
1	Conve	ersion of phenol to sali	cylaldehyde inv	olves a ———— as electrophile.
		Carbene.	(b)	Carbocation.
	(c)	Carbanion.	(d)	None of the above.
2	Reform	matsky reaction provi	des a method for	r the preparation of ————.
	(a)	Ketones.	(b)	Aldehydes.
	(c)	α-hydroxy esters.	(d)	None of the above.
5	3 Whiel	n one of the following	gives Cannizaro	's reaction ?
	(a)		(b)	$\mathrm{CH_{2}CH_{2}CHO}.$
	(c)		(d)	[CH <sub>3</sub> ] <sub>3</sub> C-CHO.
	4 Grign	nard reagent reacts wi	th methanal fol	lowed by hydrolysis gives a ———.
	(a)			Tertiary alcohol.
	(c)	Secondary alcohol.	(d)	Mixture of the above.
		h is most acidic amon	g the following?	
		m-Nitrophenol.	(b)	o-Nitrophenol.
		p-Nitrophenol.	· (d)	o-Cresol.
		n anisole is treated wi	th HI, the produ	acts are ———
				HCl, the products is ———.
	8 In re	actions, the carbonyl	carbon of ketone	es are mostly attacked by ———.
	(a)		(b)	
	(c		(d)	All of the above.
	9 Perio	cyclic reactions genera	ally involves :	

(a) Carbocations.

(c) Free radicals.

(b) Carbanions.

(d) No intermediates.

			4		Da
10	Prepa final	aration of ethers by stage.	alkoxy-mercurat	ion involves ———	- as a reducing agent
	(a)	LiAlH <sub>4</sub> .	(b)	Na BH <sub>4</sub> .	
	(c)	Ni-H <sub>2</sub> .	(d)	Pd-H <sub>2</sub> .	
11.	Alkyl	lithium in excess rea	act with CO2 follo	wed by hydrolysis gi	ves ——
		Ketone.	(b)	Carboxylic acid.	
	(c)	Aldehyde.	(d)	Diol.	
12	Which	one of the following	decolourises bro	mine water —	
		Benzoic acid.	(b)	Cinnamic acid.	
	(c)	Malonic acid.	(d)	Citric acid.	
Shor	rt Anen	vor Trongt			(12 × ¼ = 3 weight

- rt Answer Type questions. Answer all nine questions :
  - 13 Ethanol boils at a higher temperature than dimethyl ether. Why?
  - 14 Explain Saytzeff rule.
  - 15 Give any synthetic application of Reformatsky reaction.
  - 16 What is Caprolactam?
  - 17 Give an example of a pericyclic reaction taking place in human body.
  - 18 Draw the structure of aspirin.
  - 19 Why is phenolphthalein colourless in strong alkali?
  - 20 Why is it difficult to prepare Grignard reagent from allyl bromide?
  - 21 Mention a synthetic application of ethylene oxide.

 $(9 \times 1 = 9 \text{ weight})$ 

- III. Short essays or paragraph questions. Answer any five questions:
  - 22 Discuss the stereochemistry of  $S_N^2$  reaction.
  - 23 Explain the mechanism of Cannizaro's reaction.
  - 24 Give a comparative study between acetone and acetaldehyde.
  - 25 How are pericyclic reactions classified? Give one example of each.
  - 26 Discuss Diels-Alder reaction using FMO method.
  - 27 Discuss the mechanism of HVZ reaction.
  - 28 Discuss the bimolecular displacement mechanism of Chlorobenzene using alkali.

 $(5 \times 2 = 10 \text{ weights})$ 

## Essay questions. Answer any two questions:

- 29 Discuss the mechanism of the following reactions:-
  - (a) Pinacol-Pinacolone rearrangement;
  - (b) Benzoin condensation:
  - (c) Claisen-Schmidt reaction;
  - (d) Saponification of ester.
- 30 Give the effects of substituents on the acidity of:
  - (a) Aliphatic carboxylic acids;
  - (b) Phenols.
- 31 How are the following compounds prepared?
  - (a) Vanillin;
  - (b) Organocopper compounds;
  - (c) Malonic acid;
  - (d) Cinnamic acid.

 $(2 \times 4 = 8 \text{ weightage})$