ex.	10	10	0
193	411	42	1
	T ()		Acres .

Time: Three Hours

13 What is an ambidentate ligand?

14 Write the IUPAC name of [CoCl(NH<sub>3</sub>)<sub>5</sub>]<sup>2+</sup>.

(Pages 2)

Name	
Reg. No	****************

Maximum: 30 Weightage

## SIXTH SEMESTER B.Sc. DEGREE EXAMINATION, MARCH 2013

(CCSS)

## Chemistry

## CH 6B 15-INORGANIC CHEMISTRY-II

I.		ewer all the <i>twelve</i> questions. Each qualitiple choice, fill in the blanks and one			4. This section contains		
	1	Write an example for hexadentate lig-					
	2	The oxidation number of Fe in K <sub>4</sub> [Fe(CN) <sub>6</sub> ] is ———.					
	3	K <sub>3</sub> [CoF <sub>6</sub> ] is high spin complex. The hybrid state of Co atom in the complex is:					
		(a) sp <sub>3</sub> d.	(b)	$\mathrm{sp}^3\mathrm{d}^2$ .			
		(c) d <sup>2</sup> sp <sup>3</sup> .	(d)	dsp <sup>2</sup> .			
	4	In the complex Fe(CO)x, the value of	x is -	<del>-</del>			
5 Out of the following metals, which forms polynuclear carbonyl?							
		(a) Na.	(b)	Mg.			
		(c) Mn.	(d)	Al.			
	6	The colour of [Ti(H2O)6]3+ is due to —	¥ 4				
	. 7	The basic unit of porphyrine is ———	<u> </u>				
8 Carbon nanotubes were first found in the cathode deposits obtained in the arc evapora-							
				Y - 1			
	9	Write the preparation of $S_2N_2$ .					
10 Complete the following equation : $2CaO + SiO_2 \rightarrow$ .							
- 4	11	What is the substance used to slowdown the setting of cement?					
	12	Write the composition of soda glass.					
				1	$(12 \times \frac{1}{4} = 3 \text{ weightage})$		
П.	An	swer all the nine questions. Each ques	tion c	carries 1 weightage :			

- 15 Write the molecular geometry and hybridized state of [Ni(CN)4]2-.
- 16 What is Zeigler-Natta catalyst?
- 17 Describe myoglobin.
- 18 What are cytochromes?
- 19 How will you prepare boron nitride nanotubes?
- 20 Illustrate the preparation of  $\mathrm{Si}_{3}\mathrm{N}_{4}$  nanowires.
- 21 What is annealing?

 $(9 \times 1 = 9 \text{ weighta})$ 

- III. Answer any five questions. Each question carries 2 weightage:
  - 22 Draw the structure of complex [SbF<sub>5</sub>]<sup>2-</sup> and write the hybridization and geometry.
  - 23 Draw and explain the crystal field splitting in an octahedral complex.
  - 24 What are sandwich compounds? Explain.
  - 25 Describe the applications of organo arsenic compounds in medicine.
  - 26 Explain Sodium/Potassium pump .
  - 27 Describe the applications of nanotechnology in catalysis.
  - 28 Write a note on phosphate fertilizer.

 $(5 \times 2 = 10 \text{ weightag})$ 

- IV. Answer any two questions. Each question carries 4 weightage:
  - 29 Explain the structural isomerism in co-ordination compounds.
  - 30 Illustrate the preparation, properties and structure of  $S_4N_4$ .
  - 31 Describe the manufacture of Portland cement.

 $(2 \times 4 = 8 \text{ weightage})$