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SIXTH SEMESTER B.Sc. DEGREE EXAMINATION, MARCH 2012

(CCSS)

	Chemist	ry—Cor	e Course				
	CH 6B 15—Core Course V	III—IN	ORGANIC CHEMIST	RY—II			
me : Thre				Maximum: 30 Weighta			
		Section	A .v				
I. Ans	wer all questions:						
1	The name of the complex Na ₂ [SiF6	l is:					
	(a) Sodium tetrafluorosilicate (
	(b) Sodium hexafluorosilicate ((V), -					
	(c) Sodium difluorosilicate (VI)						
	(d) None of the above.						
2	Which of the following is an examp	le for hi	gh spin Octahedral com	plex?			
	(a) $[Fe(H_2O)_6]^{3+}$.		$[\mathrm{Fe}(\mathrm{CN})_6]^4$.				
	(c) [Co(NH ₃) ₆] ³⁺ .	(d)	[Fe(CN) ₆] ⁴ .				
3	Cs+CH is an example for :						
	(a) Ionic compound.	(b)	Ionic organo metallic	compound.			
	(c) Transition compound.	(d)	Ylides.				
4	The glass used in automobiles and	aeropla	nes is:				
	(a) Safety glass.		Ground glass.				
	(c) Crooke's glass.	(d)	Crown glass.				
5	Si is a nutrient in biolo	gical sys	stem.				
6	The football shaped cage like structures of carbon atoms are called ———.						
7	Phosphazenes are cyclic or chain polymers which contain ——— repeating units.						
8	A fertilizer which contains more than one of the major nutrients is called a ———.						
9	Name the complex [Co(NH ₃) ₅ CO ₃]Cl.						
10	What is the Hybridization in [Ni(C						

- 11 Give one application of Zeigler-Natta catalyst.
- 12 Name one biologically important cobalt containing compound.

 $(12 \times \frac{1}{4} = 3 \text{ weights})$

Section B

- II. Short Answer type questions. Answer all nine questions:
 - 13 What is a Chelate?
 - 14 Why does NH3 readily form complexes while NH4+ does not?
 - 15 Write down the structure of Co2(CO)8.
 - 16 What is meant by Bohr's effect?
 - 17 Name any two scanning probe instruments.
 - 18 Give any two examples for orthosilicates.
 - 19 What are ceramics? Give one use.
 - 20 What is Carborandum? What is its use?
 - 21 Give equation for Vilsmeir reaction.

 $(9 \times 1 = 9 \text{ weights})$

. Section C

- III. Short paragraph questions. Answer any five questions:
 - 22 Give the postulates of Werner's co-ordination theory.
 - 23 How is CFT useful in explaining the colour of transition metal complex?
 - 24 Explain the application of Wilkinson catalyst.
 - 25 Give short note on the biochemistry of Mg.
 - 26 Briefly explain the application of nanotechnology in biology.
 - 27 Write short note on zeolites.
 - 28 Explain the role of selenium in xerography.

 $(5 \times 2 = 10 \text{ weights})$

Section D

- IV. Essay questions. Answer any two questions:
 - 29 Describe method for the manufacture of glass. What is annealing?
 - 30 Explain qualitatively the bonding in ferrocenes. What are its properties?
 - 31 Explain the different kinds of structural isomerism possible in complexes.

 $(2 \times 4 = 8 \text{ weights})$